In Vitro Quality Evaluation of Amoxicillin Trihydrate Capsules Marketed in Gaza Strip-Palestine

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ABSTRACT

The objective of this Post-Marketing Pharmaceutical Quality Evaluation of Amoxicillin trihydrate 500mg capsules to evaluate the quality standards of seven different marketed brands with various price ranges, collected from retail drug stores of Gaza-strip, Palestine. The quality of amoxicillin trihydrate capsules was assessed through evaluation of identification, uniformity of weight, disintegration, dissolution and assay of content of active ingredient using spectrophotometric method. It was observed that six of seven brands of amoxicillin trihydrate capsules meet quality specifications in Pharmacopoeia. The spectrophotometric method which used in assay of content of active ingredient of amoxicillin trihydrate capsules is simple, inexpensive, easy to use and could be used in routine monitoring especially in the absence of high technology equipment. This study shows the needing for more market monitoring of all available brands of all drugs in the drug market of Gaza Strip-Palestine.

Keywords: Amoxicillin trihydrate, pharmaceutical quality evaluation, dissolution test, gaza strip, pharmacopoeial specifications.

INTRODUCTION

Currently, it is estimated that 10–15% of the global drugs supplied are counterfeit. The prevalence is higher in developing countries, in Africa, and in parts of Asia and Latin America where up to 30–60% of drugs on the market are counterfeit. Among the medicines, antibiotics account for 28% of global counterfeit medicines. These problems have resulted in a weak therapeutic efficiency and devel-

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opment of dire resistant strains. There is, therefore, a need to routinely assess the pharmaceutical quality of drugs¹. Amoxicillin is an oral semi-synthetic, β -lactam antibiotic used to treat bacterial infections caused by susceptible microorganisms. It is usually the drug of choice within the class because it is better absorbed, following oral administration, then other β -lactam antibiotics². Amoxicillin (500 mg capsules or tablets) is categorized as a biopharmaceutics classification system (BCS) class 1 drug. A recently published biowaiver monograph on amoxicillin trihydrate recommends submission of either comparative in vitro dissolution data or in vivo bioequivalence data as evidence to establish therapeutic equivalence of generic solid oral amoxicillin products of 250 and 500 mg³.

Post-Marketing monitoring of medicines has been performed to evaluate the quality of marketed pharmaceutical brands⁴. Quality of the drug according to the modern definition requires that the product contain the quantity of each active ingredient claimed on its label within the applicable limits of its specifications, contain the same quantity of active ingredient from one dosage unit to the next, be free from extraneous substances, maintain its potency, therapeutic availability and appearance until used, and upon administration release active ingredient for full biological availability⁵. Quality control is the part of Good Manufacture Practice (GMP) that is concerned with sampling, specifications, testing, documentation and release procedures which ensure that the necessary and relevant tests are actually carried out and that the materials are not released for use, not products released for sale or supply, until their quality has been judged to be satisfactory6. The safety and efficacy of a pharmaceutical dosage form can be guaranteed when its quality is reliable. The efficacy of pharmaceutical dosage forms generally depends on their formulation properties, and manufacturing methods, hence it is likely that the quality of dosage form may vary⁷.

The increase in the number of generic drug products from different multiple sources has placed people and prescribers in a position of selecting one from among several seemingly equivalent products⁸. Many of these products are inexpensive and affordable, but with uncertainly about their quality⁹. Several studies showed that switching from branded to generic medicine might result in changes of pharmacokinetics/pharmacodynamics profile, leading to sub-therapeutic concentration or therapeutic failure and or adverse reactions¹⁰. It is very essential to do bioequivalence studies for generic products on account of any significant difference in the rate and extent by which the therapeutic ingredients become available at the site of drug action, administered under uniform conditions in an adequately designed study¹¹. To identify bioavailability prob-

lems dissolution testing serves as an indicator¹². Biopharmaceutically as well as chemically equivalent drug products must have the same quality, strength, purity, content uniformity, disintegration and dissolution rates¹³. In vitro quality control (QC) of pharmaceutical products is a fixed set of investigation started during production by in-process quality control tests and after production by finished product quality control tests as per official pharmacopoeias and different regulatory agencies. QC tests help in avoiding the confusion regarding safety, potency, efficacy and stability of pharmaceuticals¹⁴.

Regular control of drug products has long been an integral part of the pre-and post-marketing quality control to safeguard the public. Many developing countries do not have an effective means of monitoring the quality of generic drug products in the market. This results in a widespread distribution of substandard and/or counterfeit drug products⁸.

There are a number of cases related to substandard and counterfeit drugs. Composition and ingredients of substandard drugs don't meet the correct scientific specifications for these reasons they are ineffective and often dangerous to the patient. Counterfeit drugs may include products with the correct ingredients but fake packaging, with the wrong ingredients, without active ingredients or with insufficient active ingredients¹⁵. Substandard and counterfeit drugs are a major cause of morbidity, mortality and loss of public confidence in drugs and health structures¹⁶. WHO has estimated that approximately 10% of the global pharmaceuticals market consists of counterfeit drugs, but this estimate increases to 25% in developing countries, and may exceed 50% in certain countries. FDA estimates that up to 25% of the drugs consumed in poor countries are substandard or counterfeit¹⁵. Substandard and counterfeit drugs are not only limited to poor and developing countries but also intensely noticeable in developed countries. In 2007–2008, due to the uses of adulterated blood thinner, heparin 149 Americans were dying that was legally imported. In 2012, contaminated steroids killed 11 people and sickened another 100 people in the US. In another case, vials of the cancer medicine, Avastin were found to contain no active ingredients17. In a study of WHO found that 28% of antibiotic and 20–90% of antimalarial drugs were failed quality specifications¹⁸.

As amoxicillin is widely used antibiotic in Gaza Strip, the objective of this study was to assess the quality of different leading brands of amoxicillin trihydrate 500mg capsules formulation commercially available in the market of Gaza Strip. This study also used and validated an analytical method for the assay of content of amoxicillin trihydrate in the capsules, which will be easy to use, accurate, simple and inexpensive when compared with other methods.

METHODOLOGY

Materials

Amoxicillin trihydrate (Merck, Germany), Amoxicillin trihydrate capsules (500 mg): seven brands, Sodium hydroxide, Hydrochloric acid, Ferric sulphate (Merck, Germany).

Instruments

Analytical balance (YMC, Japan), Disintegration apparatus (Toyama sangyo, Japan), Dissolution test apparatus (Apparatus II "Paddle apparatus"), Electricheating distilling apparatus, Magnetic stirrer (Heidolph, Germany), Micrometer (Mitutoyo, Japan), Micropipette (Nichiryo, Japan), UV-visible spectrophotometer (equipped with 1cm Shimadzu, 1601, matched quartz cells).

Methods

Identification test of active substance

Amoxicillin trihydrate capsules

20 capsules from each brand were weighed. A quantity of the powder from each brand containing 0.05 g of amoxicillin trihydrate was weighed, then 10 ml of 1 % ferric sulphate was added¹⁹.

Amoxicillin trihydrate pure powder

A 0.05 g of a moxicillin trihydrate pure powder was weighed, then 10 ml of 1 % ferric sulphate was added.

Uniformity of weight determination

20 capsules from each brand were taken at random and brushed from dust using soft brush then were weighed individually using analytical balance. Each capsule was opened without losing any part of the shell and the contents was removed as completely as possible then the shell was weighed. The weight of the content is the difference between weighing's (weight of capsule "content & shell" – weight of shell). The average weight of content, percentage deviation from the average weight and SD were calculated (BP 2018).

Disintegration test

Disintegration time of six units per brand was determined in distilled water at $37 \pm 1^{\circ}$ C using disintegration apparatus (BP 2018). Determination was done in triplicate, and then the mean and SD were calculated.

Dissolution test

According to official monograph, the dissolution was performed according USP 41 NF20. The dissolution rate was determined by using dissolution apparatus II, and 900 ml of distilled water. Six units were used from each brand. The dissolution medium was maintained at 37 ± 0.5 °C, and the paddle was rotated at 75 rpm. Samples (10 ml) were withdrawn at different time intervals (10, 20, 30, 45, and 60 minutes). The samples were filtered and diluted appropriately with distilled water. The absorbance was measured using UV-visible spectrophotometer at 272 nm. The content of amoxicillin trihydrate capsules in each sample was determined based on the calibration curve and regression equation which was generated according to the following procedure:

- Accurately 100 mg of pure amoxicillin trihydrate powder was dissolved in distilled water and diluted with distilled water to the mark in a 50 ml volumetric flask.
- The resulting solution was filtered then different volumes were taken (1 ml, 2 ml, 3 ml, 4 ml and 5 ml) using pipette and dilution was done with the distilled water to the mark 50 ml volumetric flask.
- Different aliquots of standard solutions (40 $\mu g/ml,$ 80 $\mu g/ml,$ 120 $\mu g/ml,$ 160 $\mu g/ml$ and 200 $\mu g/ml)$ were prepared.
- The absorbance of the resulting solutions was measured at the maximum 272 nm using UV-visible spectrophotometer.
- Procedure was done in triplicate, and the mean of the absorbance values was calculated.

A linear plot with concentration of amoxicillin trihydrate on X-axis and absorbance on Y-axis, and the regression equation were obtained using Microsoft excel. The dissolution profiles of the different brands of amoxicillin trihydrate capsules were generated from the graph of the percentage amount of the dissolved drug versus time.

Analytical method validation

The quantitative determination method for amoxicillin trihydrate in the capsules was validated through assessment of accuracy and precision. The accuracy and precision of the methods were assessed by performing recovery experiments. Recovery experiment was performed as following: to a fixed amount of drug in the dosage form (pre-analyzed), pure drug was added at three amount levels (each added amount was performed in triplicate), then total amount was found by Dibbern et al. method and % recovery of pure drug was calculated²⁰. The detailed procedures were done as the following:

Standard preparing: A quantity of pure powder containing 22.5 mg of amoxicillin trihydrate was added to 50 ml volumetric flask then 0.1 M sodium hydroxide was added till reach the mark and shaken. The resulting solution (a) was filtered and 1 ml of the filtrate was diluted to 100 ml with 0.1 M sodium hydroxide to get solution (b) with concentration of 4.5 µg/ml of amoxicillin. The absorbance of the solution (b) was measured at the maximum at 247 nm using UV-visible spectrophotometer. The content was calculated taking 286 as the value of A (1 %, 1 cm). Volumes that contain (10 µg, 20 µg and 40 µg) were taken from the solution (b).

Sample preparing: 20 capsules were taken, the contents was removed as completely as possible and weighed accurately. A quantity of the powder from capsules containing 22.5 mg of amoxicillin trihydrate was added to 50 ml volumetric flask then 0.1 M sodium hydroxide was added till reach the mark and shaken. The resulting solution (c) was filtered and 1 ml of the filtrate was diluted to 50 ml with 0.1 M sodium hydroxide to get solution (d) with concentration of 9 μ g/ml of amoxicillin. Suitable dilution was made to get solution (e) with concentration of 1.5 μ g/m of amoxicillin. The absorbance of the solution (e) was measured at the maximum at 247 nm using UV-visible spectrophotometer. The content was calculated taking 286 as the value of A (1 %, 1 cm). Volume that contain 20 μ g of drug was taken from solution (e).

Recovery experiment

After preparing the solutions of pure drug and sample of amoxicillin trihydrate the following steps were done for recovery:

- Each volume that contains (10 μ g, 20 μ g and 40 μ g) of pure drug was taken and added to volume of sample containing 20 μ g.
- The absorbance of the resulting solutions were measured using UV-visible spectrophotometer at the maximum at 247 nm.
- The total amount was calculated taking the value of A (1 %, 1 cm.) as 286 for amoxicillin trihydrate.
- The percentage recovery of pure drug was calculated as the following:

% recovery = total amount (pure drug & sample) – amount of sample/ amount of pure drug x 100

- Each added amount of pure drug to the sample was performed in triplicate.
- The mean, SD and Relative Standard Deviation % (RSD %) were calculated.

Assay of content of active ingredient

Application of validated method

20 capsules were taken, the contents was removed as completely as possible and weighed accurately. A quantity of the powder containing 22.5 mg of amoxicillin trihydrate was added to 50 ml volumetric flask then 0.1 M sodium hydroxide was added till reach the mark and shaken. The resulting solution was filtered and 1 ml of the filtrate was diluted to 50 ml with 0.1 M sodium hydroxide. The absorbance of the resulting solution was measured at the maximum at 247 nm using UV-visible spectrophotometer. The content was calculated taking 286 as the value of A $(1 \%, 1 \text{ cm})^{21}$. Determination was done in triplicate, and then the mean of percentage content and SD were calculated.

RESULTS AND DISCUSSION

Identification test of amoxicillin trihydrate

The identification test of the standard (pure amoxicillin trihydrate) and the various brands resulted as intense yellow color was produced in standard as well as in all brands. From the results it was observed that all brands contain the needed active substance by comparing the result with that of standard pure active substance.

The results of the physicochemical properties of the various brands of amoxicillin trihydrate capsules are presented in Table 1.

0.597 ± 0.016	6.363 ± 0.172	90.956 ± 0.707	108.970 ± 0.586
0.600 ± 0.005	4.630 ± 0.312	89.471 ± 1.179	107.390 ± 1.088
0.622 ± 0.005	5.210 ± 0.518	94.906 ± 0.926	110.436 ± 0.391
0.571 ± 0.019	3.863 ± 0.551	83.684 ± 1.345	107.729 ± 0.391
0.587 ± 0.007	3.153 ± 0.252	92.238 ± 0.728	109.082 ± 0.391
0.636 ± 0.015	4.473 ± 0.423	97.967 ± 0.192	107.390 ± 0.195
0.593 ± 0.006	3.517 ± 0.180	91.651 ± 0.870	107.842 ± 0.517
	$\begin{array}{c} 0.622 \pm 0.005 \\ 0.571 \pm 0.019 \\ 0.587 \pm 0.007 \\ 0.636 \pm 0.015 \end{array}$	$\begin{array}{cccc} 0.622 \pm 0.005 & 5.210 \pm 0.518 \\ 0.571 \pm 0.019 & 3.863 \pm 0.551 \\ 0.587 \pm 0.007 & 3.153 \pm 0.252 \\ 0.636 \pm 0.015 & 4.473 \pm 0.423 \end{array}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

Uniformity of weight determination

Weight variation is important to ensure good manufacturing practices (GMP) sustained by the manufacturers and the content uniformity of the formulation²². From results (Table 1) it was noticed that the uniformity of weight determination for all the brands showed compliance with the BP 2018 specification, as none of the brands deviated by up to \pm 5.0 % from their mean values. This indicates that the factors leading to weight variation were taken in consideration. Factors that affect tablet weight includes tooling of the compression machine, head pressure, machine speed and flow properties of the powder²³.

Disintegration test

It was observed that all brands of amoxicillin trihydrate capsules passed BP 2018 specification of disintegration test, as the disintegration time of amoxicillin trihydrate capsules was less than 30 minutes (Table 1).

Dissolution test

The dissolution test is the measurement of the proportion of drug dissolving in a stated time under standardized conditions in vitro²⁴. The importance of the test is to ensure the availability of the drug for absorption and to predict in vivo bioavailability⁹.

The calibration curve was shown in Figure 1, and the resulted regression equation as following:

0.600 0.500 y = 0.0028x + 0.0092 $R^2 = 0.999998$ 0.400 Absorbance 0.300 0.200 0.100 0.000 0 40 80 120 160 200 240 Concentration of amoxicillin trihydrate in distilled water (µg / ml)

 $Y = 0.0028 x + 0.0092, R^2 = 0.999998$

Figure 1: Calibration curve for amoxicillin trihydrate in distilled water.

The USP 41 stipulated that at 60 minutes, all capsules should have released into the dissolution medium an amount not less than 80.0% of the labeled amount of amoxicillin trihydrate. The percentage mean of the amount released at 60 minutes which are represented in Table 1 showed that all brands passed the dissolution test, that all brands released more than 80.0% of their content within 60 minutes. It was observed that all brands meet pharmacopoeia specification of dissolution test. The results revealed that all brands exhibit good release of the drug to the site of absorption and may have good bioavailability. It is interesting to note that several authors have previously disagreed on the correlation between disintegration time and dissolution time. Some authors mention that disintegration and dissolution times are correlated, while others continue to disagree⁹. From the results, it was observed that no high range between times of the disintegration of brands, but it was noticed the differences in the dissolution profiles between brands as shown in Figure 2. Dissolution of drugs can be influenced by the physicochemical properties of the drug substance, the dosage form design, the manufacturing process, and the testing conditions²⁵. As there are many factors affecting on the dissolution, this gives each product certain dissolution characteristics which varies from brand to another. So it is not surprising to observe variation in vitro dissolution among seven brands amoxicillin trihydrate capsules which were investigated in this study.



Figure 2: Dissolution profile of the seven different brands of amoxicillin trihydrate capsules in distilled water.

Analytical method validation

From the results (Table 2) it was observed that the recoveries percentage of the added pure amoxicillin trihydrate were in the range of (98.506 - 102.593 %), this indicated excellent accuracies and no interference from excipients was exhibited. SD is less than 2.0 and RSD % is less than 2.0 %, this indicated the high precision of the method.

	Amount of amoxicillin trihydrate in formulation (µg)	Amount of pure amoxicillin trihydrate added (µg)	% Recovery of pure amoxicillin trihydrate*	SD	RSD %
A1	20.0	10.0	99.369	0.865	0.870
	20.0	20.0	101.249	0.330	0.326
	20.0	40.0	100.023	0.804	0.804
A2	20.0	10.0	99.952	0.158	0.158
	20.0	20.0	101.756	1.116	1.096
	20.0	40.0	99.577	0.000	0.000
A3	20.0	10.0	100.576	0.000	0.000
	20.0	20.0	101.118	1.175	1.162
	20.0	40.0	98.506	0.000	0.000
A4	20.0	10.0	100.570	0.000	0.000
	20.0	20.0	102.593	1.167	1.137
	20.0	40.0	99.886	0.819	0.820
A5	20.0	10.0	99.136	1.040	1.050
Ab				1.842	1.858
	20.0	20.0 40.0	101.054	0.000	0.000
A6	20.0	10.0	100.424	0.000	0.000
	20.0	20.0	100.362	0.917	0.913
	20.0	40.0	100.608	0.845	0.839
A7	20.0	10.0	101.049	0.000	0.000
	20.0	20.0	101.152	1.897	1.876
	20.0	40.0	99.810	0.927	0.929

Table 2. Results of recovery	study of amoxicillin trihydrate	using Dibbern et al. method

*mean value of three determinations.

Thus, the Dibbern et al²¹. method is simple, rapid, no laborious time consuming, inexpensive and no need for high cost instruments. The significant advantage is the possibility of using the method to assay the drug in complex dosage formulation in presence of the excipient without any interferences.

Assay of content of amoxicillin trihydrate

Assay of pharmaceutical products is a critical quality parameter required to confirm that the labeled amount of drug is available in a given dosage form and failure to meet the standard will result in poor quality medicines. Inadequate amounts of active pharmaceutical ingredient (API) will result in under-dosed medication, leading to poor treatment outcomes while excessive amounts of API cause over-dosage of medication, leading to increased adverse drug reactions and treatment failure²⁶.

The results showed that brands (A1, A2, A4, A5, A6, and A7) had values range (107.390-109.082% w/w), thus it lies within BP 2018 acceptable range (92.5-110.0% w/w), while brand (A3) had value 110.436 % w/w, thus not lies in BP 2018 acceptance range (Table 1).

It was observed that one of seven brands of amoxicillin trihydrate capsules (A3) failed to be within BP 2018 specification range of (92.5-110.0% w/w). This revealed that there is a problem in manufacturing of failed brands, while there is good manufacturing for accepted brands.

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