

SYNTHESIS AND CHARACTERIZATION OF SOME 2-(SUBSTITUTED  
ARYLACYLTHIO)THIAZOLINE DERIVATIVES

BAZI 2-(SÜBSTİTÜE ARILTİYO)TİYAZOLİN TÜREVLERİNİN SENTEZİ VE YAPI TAYİNİ

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In this study, five new 2-(substituted arylacylthio)thiazoline derivatives were synthesized by reacting 2-thiazoline-2-thiol with arylacylhalides. The structures of the compounds obtained were elucidated by IR, <sup>1</sup>H-NMR, FAB-MASS spectral data and elementary analyses results.

Bu çalışmada, 2-tiyazolin-2-tiyol ile arilhalojenür türevleri reaksiyona sokularak, beş yeni 2-(sübstütüe arilasiltiyo)tiyazolin türevleri sentezlendi. Elde edilen bileşiklerin yapıları, IR, <sup>1</sup>H-NMR ve FAB-MS spektroskopik verileri ve elemental analiz sonuçları yardımı ile aydınlatıldı.

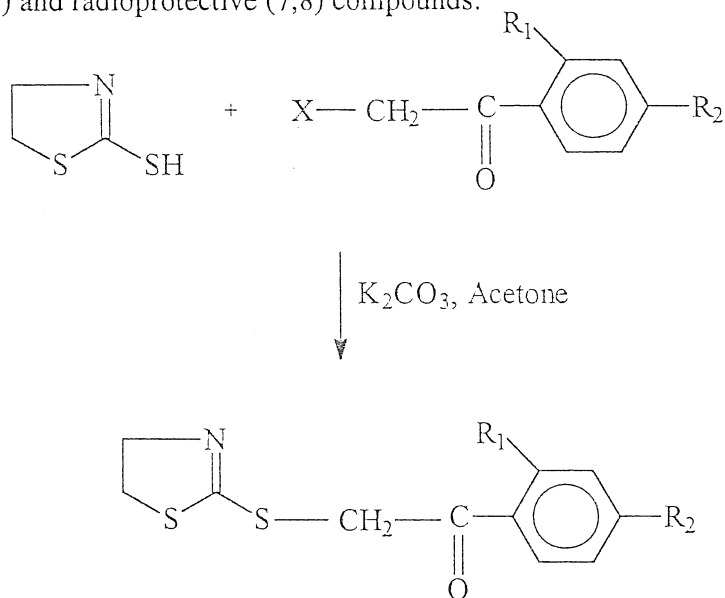
**Keywords:** 2-Thiazoline-2-thiol;  $\alpha$ -bromoacetophenone; 2-( $\alpha$ -chloroacetyl)phenol; 2-(substituted arylacylthio)thiazoline

**Anahtar kelimeler:** 2 - Tiyazolin - 2 - tiyol;  $\alpha$ -bromoasetofenon; 2-( $\alpha$ -kloroasetil)fenol; 2-(sübstütüe arilasiltiyo)tiyazolin

### Introduction

2-Thiazoline-2-thiole derivatives are known as potent insecticide (1), antithyroid (2-5), competitive inhibitor of acetylcholinesterase (6) and radioprotective (7,8) compounds.

In this work, we report the synthesis and characterization of 2-(substituted arylacylthio)thiazoline derivatives (Figure).



X = Cl, Br

R<sub>1</sub> = H, OH  
Figure

R<sub>2</sub> = H, Cl, NO<sub>2</sub>, CH<sub>3</sub>, OCH<sub>3</sub>

Correspondance

## Materials and Methods

Melting points were determined using a Gallenkamp apparatus and are uncorrected. The IR spectra were recorded in potassium bromide discs on a Shimadzu IR-435 Spectrophotometer. The  $^1\text{H-NMR}$  spectra were obtained by Bruker 250 MHz and Jeol 90 MHz in  $\text{DMSO-d}_6$  using TMS as the internal standard.

FAB<sup>+</sup>-MS were recorded at the University of Montpellier I, France. Microanalytical datas (C,H,N) agreed with the proposed structures within  $\pm 0.4\%$  of the theoretical values and were carried out by the Micro-analytical Section of Service Central (CNRS, Ecole Normale Chimie de Montpellier, France).

### Synthesis of the compounds

General procedure of 2-(arylacylthio)thiazoline derivatives

A mixture of the appropriate  $\alpha$ -bromoacetophenone (9-11) (3 mmol) [or 2-( $\alpha$ -chloroacetyl)phenol derivatives (12-14)], 2-thiazoline-2-thiol (3 mmol) and  $\text{K}_2\text{CO}_3$  (3 mmol) in acetone was stirred at room temperature for 3 hrs (Figure). The mixture was filtered, the filtrate was evaporated until dryness. Residue was recrystallized from ethanol. Some characteristics of the compounds are given in Table 1.

#### 2-(Phenacylthio)thiazoline 1

This compound was obtained by the reaction of  $\alpha$ -bromoacetophenone and 2-thiazoline-2-thiol (15)

IR (KBr,  $\text{cm}^{-1}$ ); 2911 (C-H), 1691 (C=O), 1613-1468 (C=N, C=C).  $^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ )  $\delta$  ppm; 3.50 (2H, t,  $\text{C}_5$  protons of thiazoline), 4.05 (2H, t,  $\text{C}_4$  protons of thiazoline), 4.85 (2H, s,  $\text{COCH}_2$ ), 7.50-7.75 (3H, m,  $\text{C}_3$ ,  $\text{C}_4$ ,  $\text{C}_5$  aromatic protons), 8.05 (2H, d  $J=8.34$  Hz,  $\text{C}_2$  and  $\text{C}_6$  aromatic protons). FAB<sup>+</sup> MS  $M+1$ :  $m/z$ : 238.

#### 2-(4-Chlorophenacylthio)thiazoline 2

This compound was obtained by the reaction of  $\alpha$ -bromo-4-chloroacetophenone and 2-thiazoline-2-thiol.

IR (KBr,  $\text{cm}^{-1}$ ); 2959 (C-H), 1696 (C=O), 1500-1449 (C=N, C=C).  $^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ )  $\delta$  ppm; 3.50 (2H, t,  $\text{C}_5$  protons of thiazoline), 4.10 (2H, t,  $\text{C}_4$  protons of thiazoline), 4.85 (2H, s,  $\text{COCH}_2$ ), 7.65 (2H, dd  $J=7.21$  Hz and  $J=1.80$ , 1.75 Hz  $\text{C}_3$ ,  $\text{C}_5$  aromatic protons), 8.05 (2H, dd  $J=6.80$  Hz, and  $J=1.81$ , 1.77 Hz  $\text{C}_2$  and  $\text{C}_6$  aromatic protons). FAB<sup>+</sup> MS  $M+1$ :  $m/z$ : 272.

#### 2-(4-Nitrophenacylthio)thiazoline 3

$\alpha$ -Bromo-4-nitroacetophenone and 2-thiazoline-2-thiol were used for the synthesis of this compound

IR (KBr,  $\text{cm}^{-1}$ ); 2999-2847 (C-H), 1685 (C=O), 1620-1518 (C=N, C=C).  $^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ )  $\delta$  ppm; 3.50 (2H, t,  $\text{C}_5$  protons of thiazoline), 4.10 (2H, t,  $\text{C}_4$  protons of thiazoline), 4.90 (2H, s,  $\text{COCH}_2$ ), 8.25 (2H,

dd  $J=6.99$  Hz and  $J=2.04$ , 1.75 Hz  $\text{C}_2$  and  $\text{C}_6$  aromatic protons), 8.40 (2H, dd  $J=6.98$  Hz, and  $J=1.95$ , 1.82 Hz  $\text{C}_3$  and  $\text{C}_5$  aromatic protons). FAB<sup>+</sup> MS  $M+1$ :  $m/z$ : 283.

#### 2-(2-Hydroxyphenacylthio)thiazoline 4

This compound was obtained by the reaction of 2-( $\alpha$ -chloroacetyl)phenol and 2-thiazoline-2-thiol

IR (KBr,  $\text{cm}^{-1}$ ); 3437 (O-H), 1629 (C=O), 1576-1433 (C=N, C=C).  $^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ ) (90 MHz)  $\delta$  ppm; 3.50 (2H, t,  $\text{C}_5$  protons of thiazoline), 4.00-4.30 (2H, t,  $\text{C}_4$  protons of thiazoline), 4.90 (2H, s,  $\text{COCH}_2$ ), 6.90-7.80 (3H, m,  $\text{C}_3$ ,  $\text{C}_4$  and  $\text{C}_5$  aromatic protons), 7.80-8.00 (1H, d  $J=8.00$  Hz,  $\text{C}_6$  aromatic protons). FAB<sup>+</sup> MS  $M+1$ :  $m/z$ : 254.

#### 2-(2-Hydroxy-4-methylphenacylthio)thiazoline 5

2-( $\alpha$ -Chloroacetyl)-5-methylphenol and 2-thiazoline-2-thiol were used for the synthesis of this compound

IR (KBr,  $\text{cm}^{-1}$ ); 3408 (O-H), 1633 (C=O), 1572-1441 (C=N, C=C).  $^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ ) (250 MHz)  $\delta$  ppm; 2.30 (3H, s,  $\text{CH}_3$ ), 3.50 (2H, t,  $\text{C}_5$  protons of thiazoline), 4.10 (2H, t,  $\text{C}_4$  protons of thiazoline), 4.80 (2H, s,  $\text{COCH}_2$ ), 6.80-6.95 (2H, m,  $\text{C}_3$  and  $\text{C}_5$  aromatic protons), 7.80 (1H, d  $J=8.12$  Hz,  $\text{C}_6$  aromatic proton), 11.4 (1H, s, OH).

#### 2-(2-Hydroxy-4-methoxyphenacylthio) thiazoline 6

2-( $\alpha$ -Chloroacetyl)-5-methoxyphenol and 2-thiazoline-2-thiol were used for the synthesis of this compound

IR (KBr,  $\text{cm}^{-1}$ ); 3482 (O-H), 1623 (C=O), 1566-1439 (C=N, C=C).  $^1\text{H-NMR}$  ( $\text{DMSO-d}_6$ ) (250 MHz)  $\delta$  ppm; 3.40 (2H, t,  $\text{C}_5$  protons of thiazoline), 3.75 (3H, s,  $\text{OCH}_3$ ), 4.05 (2H, t,  $\text{C}_4$  protons of thiazoline), 4.75 (2H, s,  $\text{COCH}_2$ ), 6.45 (1H, d  $J=2.3$  Hz,  $\text{C}_3$  aromatic proton), 6.50 (1H, dd  $J=8.90$  Hz and  $J=2.4$ , 2.4 Hz  $\text{C}_5$  aromatic proton), 7.85 (1H, d  $J=8.88$  Hz  $\text{C}_6$  aromatic proton), 11.85 (1H, s, OH).

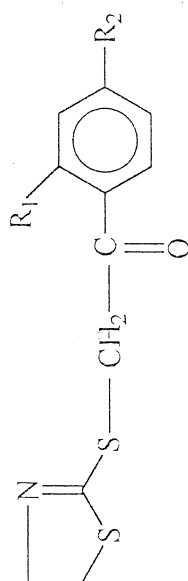
## Results and Discussion

The reaction of arylacylhalides with 2-thiazoline-2-thiol gave 2-(substituted arylacylthio)thiazoline derivatives.

The structures of the compounds were confirmed by IR,  $^1\text{H-NMR}$ , FAB<sup>+</sup>-MS spectral data and elemental analyses.

The IR spectra of the compounds 4,5,6 showed characteristic broad O-H stretching

Table 1. Some Characteristics of Compounds



No	R <sub>1</sub>	R <sub>2</sub>	M.p. °C	Yield %	Molecular Formula	Elemental Analyses calc./found		
						C	H	N
1	H	H	56	87	C <sub>11</sub> H <sub>11</sub> NO <sub>2</sub> 237.35	55.66 55.76	4.67 4.48	5.90 5.86
2	H	Cl	78	80	C <sub>11</sub> H <sub>10</sub> ClNO <sub>2</sub> 271.79	48.61 48.90	3.71 3.48	5.15 5.09
3	H	NO <sub>2</sub>	88	86	C <sub>11</sub> H <sub>10</sub> N <sub>2</sub> O <sub>3</sub> S <sub>2</sub> 282.34	46.79 46.73	3.57 3.32	9.92 9.56
4	OH	H	122	72	C <sub>11</sub> H <sub>11</sub> NO <sub>2</sub> S <sub>2</sub> 253.35	52.14 51.83	4.37 4.24	5.52 5.45
5	OH	CH <sub>3</sub>	64	67	C <sub>12</sub> H <sub>13</sub> NO <sub>2</sub> S <sub>2</sub> 267.37	53.94 53.77	4.90 4.59	5.24 5.00
6	OH	OCH <sub>3</sub>	87	65	C <sub>12</sub> H <sub>13</sub> NO <sub>3</sub> S <sub>2</sub> 283.37	50.86 50.72	4.62 4.55	4.94 4.78

bands in the 3480-3400  $\text{cm}^{-1}$  region. The C=O stretching bands were observed at 1691, 1696 and 1685  $\text{cm}^{-1}$  for the compounds 1, 2 and 3 and at 1629, 1633 and 1623  $\text{cm}^{-1}$  for the compounds 4,5,6 respectively. The presence of the O-H group in the compounds 4,5,6 caused the stretching bands of C=O groups to shift to in lower frequencies.

In the  $^1\text{H-NMR}$  spectra of the compounds,  $\text{C}_5$  and  $\text{C}_4$  protons of thiazoline resonated at 3.40 and 4.05 ppm as triplets, respectively. The  $\text{COCH}_2$  protons were observed as a singlet at 4.80 ppm. The O-H proton showed a singlet at about 11.4 ppm. All the other protons were observed in the expected regions.

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